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# Performance analysis of an organic Rankine cycle for a solar-boosted ocean thermal energy conversion system according to working fluids

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**Abstract:** Ocean Thermal Energy Conversion is a power generation cycle that utilizes temperature differences between warm surface seawater and cool deep seawater. A study on Solar-boosted Ocean Thermal Energy Conversion (SOTEC) that utilizes not only ocean thermal energy, but also solar thermal energy as a heat source has been recently researched. In SOTEC, the temperature of surface seawater is boosted using a solar thermal collector. The working fluids and cycle configuration are important factors that affect the thermal performance of an organic Rankine cycle in SOTEC. The simple Rankine cycle and the Rankine cycle with an open feedliquid heater were considered in this study. Working fluids were selected depending upon working conditions, environment, and safety factors. Exergy analysis is an important criterion for the thermal performance evaluation of cycles. This study performed exergy analysis on cycles based on SOTEC. This analysis showed that RE245fa2 has the best thermodynamic performance among the working fluids tested. The exergy efficiency of RE245fa2 was 64.76% in the simple Rankine cycle and 67.79% in the Rankine cycle with an open feedliquid heater. In the Rankine cycle with an open feedliquid heater, an increase in thermodynamic performance could be expected in a SOTEC system, compared to the simple Rankine cycle.

Keywords: Solar-boosted ocean thermal energy conversion, Working fluid, Organic Rankine cycle, Exergy analysis

Nomenclature			: Heat sink temperature
$f_p$	: Pressure ratio	$T_R$	: Reduced temperature
h	: Enthalphy	$V_m$	: Molar volume
İ	: Irreversibility rate	Ŵ	: Power
ṁ	: Mass flow rate	η	: Efficiency
Р	: Pressure	ω	: Acentric factor
$P_{high}$	: High pressure		
$P_{liq}$	: Liquid pressure		Subscript
$P_{low}$	: Low pressure	а	: Vapour except bleeding
$P_{mid}$	: Medium pressure	b	: Bleeding vapour
$P_{vap}$	: Vapour pressure	с	: Condenser
Ż	: Heat rate	crit	: Critical
R	: Gas constant	csi	: Cooling seawater inlet
s	: Entropy	cso	: Cooling seawater outlet
Т	: Pressure	е	: Evaporator
$T_0$	: Ambient temperature	р	: Pump
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t : Turbine

th : Thermal

total : Total

wsi : Warm seawater inlet

wso : Warm seawater outlet

 $\Pi$  : Second law, Exergy

### 1. Introduction

Research shows that the environmental issues of pollution, global warming, and energy shortage are gradually increasing [1]. A new regeneration energy method is one way to solve these problems.

Ocean Thermal Energy Conversion (OTEC) is a power generation cycle that utilizes temperature differences between warm surface seawater and cool deep seawater [2]. OTEC can obtain sufficient economic feasibility in cases where the temperature difference between surface seawater and deep seawater is greater than 20°C [3]. The potential energy reserves of the ocean temperature difference are approximately 4000 times that of the annual global energy consumption [4]. Therefore, due to the benefits of sustainable clean energy, many countries such as USA and Japan have conducted continued research on OTEC [5].

A study on Solar-boosted Ocean Thermal Energy Conversion (SOTEC) that uses not only ocean thermal energy, but also solar thermal energy as a heat source has been recently researched. In a SOTEC system, the temperature of surface seawater is boosted using a solar thermal collector. Yamada et al. showed that the thermal efficiency of a proposed SOTEC plant was approximately 1.5 times higher than that of conventional OTEC plants [6].

The use of an Organic Rankine Cycle (ORC) is necessary because the temperature difference between the heat source and the heat sink is too low in OTEC systems [7]. Therefore, research into the ORC characteristics required to implement this technology is needed. Cycle configuration and working fluids will affect the cycle performance greatly. Kim et al. investigated an ORC for OTEC systems, and conducted a performance analysis of the working fluids and cycles [8].

The objectives of this study are to investigate the efficiency of various cycles, select an appropriate working fluid, define the design factors, and perform an exergy analysis on a SOTEC system.

### 2. Organic Rankine cycle for SOTEC

Figure 1 and Figure 2 show the schematic diagram and the T-s diagram of the simple Rankine cycle for SOTEC, respectively. The simple Rankine cycle is a basic cycle in the OTEC system [8]. In a SOTEC system, the surface seawater receives additional heat energy from a solar collector. It is expected that there will be an increase in the turbine inlet temperature and thermal efficiency. This study referred to the research of Yamada *et al.* [6], which defined an "indirect" SOTEC system. The basic principles and thermodynamic equations of the simple Rankine cycle are detailed in the study by Kim *et al.* [8][9].

Figure 3 and Figure 4 show the schematic diagram and the T-s diagram of the Rankine cycle with an open feedliquid heater for SOTEC, respectively. It has a regenerator called "an open feedliquid heater," that mixes the liquid discharged by a working fluid pump and bleeds gas from the turbine outlet [10]. This cycle is expected to have an increase in efficiency when compared to the simple Rankine cycle [11]-[13]. Thermodynamic states of this cycle can be expressed in Equations (1) - (13), according to the first and second laws of thermodynamics, based on energy and exergy analysis.



Figure 1: Schematic diagram of the simple Rankine cycle



Figure 2: T-s diagram of the simple Rankine cycle

Class	Dry fluid		Isentropic fluid		Wet fluid		
Substance	R236ea	RE245fa2	R134a	R1234yf	R32	R152a	R161
Туре	HFC	HFC	HFC	HFO	HFC	HFC	HFC
CAS no.	431-63-0	1885-48-9	811-97-2	754-12-1	75-10-5	75-37-6	353-36-6
$P_{vap@44^{\circ}C}[MPa]$	0.38	0.17	1.13	1.13	2.73	1.01	1.5
$P_{liq@10^{\circ}C}[MPa]$	0.12	0.05	0.41	0.44	1.11	0.37	0.6
ALT[years]	10.7	2.2	14	0.03	4.9	1.4	0.3
GWP	1,370	286	1,430	4	675	124	12
ODP	0	0	0	0	0	0	0
ASHRAE safety group	-	-	A1	A2L	A2L	A2	-

Table 1: Properties of the working



**Figure 3:** Schematic diagram of the Rankine cycle with an open feedliquid heater



Figure 4: T-s diagram of the Rankine cycle with an open feedliquid heater

$$\dot{W}_t = \dot{m}_{total}(h_1 - h_2) + \dot{m}_a(h_2 - h_3) \tag{1}$$

$$\dot{I}_t = T_0 \{ \dot{m}_{total} (s_2 - s_1) + \dot{m}_a (s_3 - s_2) \}$$
(2)

$$\dot{Q}_c = \dot{m}_a (h_3 - h_4)$$
 (3)

$$\dot{I}_c = T_0 \dot{m}_a \{ (s_4 - s_3) - (h_4 - h_3/T_L) \}$$
(4)

$$\dot{W}_p = \dot{m}_{total}(h_7 - h_6) + \dot{m}_a(h_5 - h_4)$$
(5)

$$\dot{I}_p = T_0\{\dot{m}_{total}(s_7 - s_6) + \dot{m}_a(s_5 - s_4)\}$$
(6)

$$\dot{Q}_e = \dot{m}_{total}(h_1 - h_7) \tag{7}$$

$$\dot{I}_e = T_0 \dot{m}_{total} \{ (s_1 - s_7) - (h_1 - h_7 / T_H) \}$$
(8)

$$\dot{m}_{total}h_6 = \dot{m}_a h_5 + \dot{m}_b h_2 \tag{9}$$

$$\dot{I}_r = T_0 (\dot{m}_{total} s_6 - \dot{m}_a s_5 - \dot{m}_b s_2)$$
(10)

$$\dot{I}_{total} = T_0 \{ -\dot{m}_{total} \left( h_1 - h_7 / T_H \right) - \dot{m}_a (h_4 - h_3 / T_L) \}$$
(11)

$$\eta_{th} = (\dot{W}_t - \dot{W}_p) / \dot{Q}_e \tag{12}$$

$$\eta_{\Pi} = \eta_{th} / (1 - T_L / T_H) \tag{13}$$

In the Rankine cycle with open feedliquid heaterankine cycle with an open feedliquid heater, the pressure ratio  $(f_p)$  needs to be defined because of the existence of a medium pressure  $(P_{mid})$  exists **[8][12][13]**. After the high pressure  $(P_{high})$  and low pressure  $(P_{low})$  are determined, the medium pressure  $(P_{mid})$  can be defined as follows.

$$P_{mid} = P_{low} + f_p (P_{high} - P_{low})$$
(14)

# 3. Cycle analysis condition

### 3.1 Selection of working fluids

A study by Kim *et al.* **[8][9]** was referred to in the selection of working fluids for this study. Working fluids were selected by referring to NIST Refprope. CFC and HCFC affiliated refrigerants that have caused environmental problems, such as ozone depletion and global warming, were excluded due to environmental regulations. Working fluids demanding that evaporation is above 3 MPa or critical pressure in design condition were additionally excluded. All the working fluids selected have an Atmosphere Life Time (ALT) less than 1,000 years, a Global Warming Potential (GWP) less than 1,500, and an Ozone Depletion Potential (ODP) of 0. The American Society of Heating, Refrigerating and Air-conditioning Engineers (ASHRAE) safety group of the fluids was also considered, and fluids in groups Al, A2L, and A2 were chosen as they have low toxicity and flammability. Seven working fluids, shown in Table 1, were selected. They can be classified into dry, isentropic, and wet fluid categories.

#### 3.2 Design parameters of the cycle

Pinch Point Analysis (PPA) is a thermodynamic cycle analysis method that uses Pinch Point Temperature Difference (PPTD) [14]. The PPTD ensures that the temperature and pressure in an evaporator and a condenser are constantly maintained, so that a comparison analysis of each cycle and the various working fluids can be performed. The design parameters of this study, shown in Table 2, were based on previous research that used PPA [8][9]. The PPTD of the heat exchangers is the same value as was used in the research of Aydin et al. [15]. The temperature increase in the surface seawater due to the solar collector was assumed to be 20°C, as chosen by Yamada et al. [6]. Aspen HYSYS, a thermal process and cycle design program, was used as an analysis tool. A SOTEC system was replicated using this program. The selected working fluids were applied, and the thermodynamic performance was calculated in each cycle.

Table 2: Design	parameters o	of the thermod	ynamic cy	ycle
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Parameters	Values
Surface seawater temperature passing the solar collector[°C]	48
deep seawater temperature [ $^{\circ}$ C ]	5
Seawater temperature difference between inlet and outlet at the evaporator and condenser[°C]	3
Evaporator pinch point temperature difference[°C]	2
Condenser pinch point temperature difference [ $^{\circ}C$ ]	2
Evaporation exit vapour quality	1
Condenser exit vapour quality	0
Surface seawater flow rate	100
Turbine adiabatic efficiency [%]	85
Pump adiabatic efficiency [%]	80
Ambient temperature [°C]	20

#### 3.3 Peng-Robinson equation

Since the ideal gas equation has limited use in practical applications, the real gas state equations were used to precisely indicate the state of the substance [10]. In this study, the Peng-Robinson equation was used as well as the real gas state equations, and they can be expressed as follows.

$$P = RT/(V_m - b) - a\alpha/(V_m^2 + 2bV_m - b^2)$$
(15)

$$a = 0.45724R^2 T_{crit}^2 / P_{crit}$$
(16)

$$b = 0.07780RT_{crit}/P_{crit} \tag{17}$$

$$\alpha = \{1 + \kappa (1 - T_R^{0.5})\}^2 \tag{18}$$

$$\kappa = 0.37464 + 1.54226\omega - 0.26992\omega^2 \tag{19}$$

$$T_R = T/T_{crit} \tag{20}$$

# 4. Results and discussion

#### 4.1 Simple Rankine cycle



Figure 5: Exergy efficiency of the simple Rankine cycle



Figure 6: Cycle irreversibility of the simple Rankine cycle

Figure 5 shows the exergy efficiency of the simple Rankine cycle in a SOTEC system, referring to the study by Kim *et al.* [8][9]. RE245fa2 showed the highest exergy efficiency (64.76%) and R1234yf showed the lowest exergy efficiency (62.23%). These results are similar to those of the study by Kim *et al.* [8][9] that performed thermodynamic comparison analysis of various working fluids in an OTEC system. Therefore, RE245fa2 showed the best thermodynamic performance among the selected working fluids, and has the potential to be a working fluid in a SOTEC system as well as an OTEC system.

**Figure 6** shows cycle irreversibility of the simple Rankine cycle in a SOTEC system, by referring to the study by Kim *et al.* **[8][9].** R1234yf showed the highest cycle irreversibility (62.57 kW), and RE245fa2 showed the lowest cycle irreversibility (58.24 kW). Cycle irreversibility is the difference between reversible work and useful work **[10]**. The cycle irreversibility of the working fluids was inversely proportional to exergy efficiency. The reason for this phenomenon is that the recovered exergy decreases as the exergy is destroyed, which means cycle irreversibility increases **[10]**. Exergy destruction is greatly influenced by the entropy generation of a cycle and the ambient temperature.

#### 4.2 Rankine cycle with an open feedliquid heater

Figure 7 shows exergy efficiency of the Rankine cycle with an open feedliquid heater vs. the pressure ratio in a SOTEC system. This was calculated using **Equation (13)**. All of the selected working fluids showed peak thermal performance during this cycle when the pressure ratio was 0.4. This means that the quantity of vapor bled from the turbine should provide this pressure ratio for maximum efficiency.



**Figure 7:** Exergy efficiency of the Rankine cycle with an open feedliquid heater vs. the pressure ratio

**Figure 8** shows exergy efficiency of the Rankine cycle with an open feedliquid heater, when the pressure ratio is 0.4, compared to the simple Rankine cycle in a SOTEC system. RE245fa2 showed the highest exergy efficiency (67.79%). R1234yf, which had the lowest exergy efficiency, had the greatest increase in exergy efficiency (3.78%), and R152a showed the smallest increase of 2.96%. The selected working fluids showed different thermodynamic characteristics, such as mass flow and enthalpy, depending on the design conditions.

Figure 9 shows cycle irreversibility of the Rankine cycle with an open feedliquid heater vs. the pressure ratio in a SOTEC system. This was calculated using Equation (11). All of the selected working fluids showed the lowest cycle irreversibility during this cycle when the pressure ratio was 0.4. This is because the cycle irreversibility of working fluids is inversely proportional to exergy efficiency, as mentioned previously.







**Figure 9:** Cycle irreversibility of the Rankine cycle with an open feedliquid heater vs. the pressure ratio



**Figure 10:** Cycle irreversibility of the Rankine cycle with an open feedliquid heater compared to the simple Rankine cycle

**Figure 10** shows the cycle irreversibility of the Rankine cycle with an open feedliquid heater when the pressure ratio is 0.4, compared to the simple Rankine cycle. R1234yf showed the highest cycle irreversibility (58.07 kW) and the greatest decrease in cycle irreversibility (4.50 kW). R152a showed the lowest decrease (3.17 kW). The cycle irreversibility of all the working fluids is affected by the entropy generation in this cycle.

All of the selected working fluids in this cycle showed an improvement in thermodynamic performance, as seen in the results of the studies by Kim *et al.* [8][9], Shin *et al.* [12], and Kim *et al.* [13]. Therefore, to further improve thermodynamic performance, the installation of multiple regenerators (open or closed type of feedliquid heaters) has been proposed [10].

# 5. Conclusion

In the organic Rankine cycle for a SOTEC system, RE245fa2 showed the best thermodynamic performance in this study. The exergy efficiency of RE245fa2 was 64.76% in the simple Rankine cycle and 67.79% in the Rankine cycle with an open feedliquid heater. However, there was no significant performance difference when different working fluids were used. Therefore, when selecting a working fluid not only the thermodynamic performance but also the environmental and safety factors, which are of the utmost importance, should be considered.

In the case of the Rankine cycle with an open feedliquid heater, the ideal pressure ratio with regard to the medium pressure was 0.4. The Rankine cycle with an open feedliquid heater showed cycle performance improvements when compared to the simple Rankine cycle. It can be expected that better cycle performance improvements can be achieved using additional feedliquid heaters or other regenerative design methods. Finally, future work should consider the economic feasibility of SOTEC when the hardware of an organic Rankine cycle is added.

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